

## Year 10 Chemistry Learning Programme 3

<p>The LORIC skill focus for his LP is: RESILIENCE Respect - treat others how you would wish to be treated yourself. Compassion - the quality of feeling pity and concern for the sufferings or misfortunes of others. Honesty - the quality of being truthful.</p>		<p><b>Literacy:</b></p> <ul style="list-style-type: none"> <li>Capital letters must be used at the start of sentences and for the first letter of proper nouns</li> <li>Full stops must be used at the end of a sentence</li> <li>Question marks must be used at the end of a question</li> <li>Apostrophes should only be used for possession or omission</li> <li>Days of the week and months must be spelled correctly</li> <li>Key words must be spelled correctly</li> </ul>	
<p><b>What will I be learning about in this Learning Programme?</b> How to calculate the number of moles in reactions involving solid, gases and solutions. How to complete a titration and how to identify the products of acid reactions using ionic equations.</p>			
<p><b>Where have I seen this learning before?</b> Acids at KS3</p> <p><b>What could I use it for?</b> Chemistry calculations are used across the course and on into KS5.</p>			
<b>In LP3.1, I will know:</b>	<b>06/01/25 - (WK 2)</b>	<b>Key Vocabulary</b>	<b>Homework</b>
<p>how to calculate the number of moles or mass of a substance from data supplied; how to calculate the number of moles of a substance using molar ratio and balanced symbol equations; how to use balanced symbol equations to calculate reacting masses.</p>		Mole	Retrieval Questions based on LP2
<b>In LP3.2, I will know:</b>	<b>13/01/25 - (WK 1)</b>	<b>Key Vocabulary</b>	<b>Homework</b>
<p>how to use balanced symbol equations to calculate reacting masses; how to use balanced symbol equations to calculate reacting masses when there is a limiting reactant.</p>		Limiting Reactant	PPQ on balanced equations
<b>In LP3.3, I will know:</b>	<b>20/01/25 - (WK 2)</b>	<b>Key Vocabulary</b>	<b>Homework</b>
<p>how to calculate the percentage yield using a variety of units and conversions and evaluate different reactions to decide the best production method of a chemical; how to calculate the mass of a chemical when any volume and concentration is given and independently express your answer to an appropriate number of significant figures;</p>		Percentage Yield	Retrieval Questions based on moles
<b>In LP3.4, I will know:</b>	<b>27/01/25 - (WK 1)</b>	<b>Key Vocabulary</b>	<b>Homework</b>
<p>how precise results are obtained in a titration; how to calculate the unknown concentration of a reactant in a neutralisation reaction when the volumes are known and the concentration of one reactant is also known.</p> <p>Extended Task.</p>		Concordant Results	PPQ based on quantitative chemistry so far
<b>In LP3.5, I will know:</b>	<b>03/02/25 - (WK 2)</b>	<b>Key Vocabulary</b>	<b>Homework</b>
<p>how to calculate the moles or volume of a gaseous substance involved in a chemical reaction.</p>		Volume	Retrieval Questions based on quantitative chemistry
<b>In LP3.6, I will know:</b>	<b>10/02/25 - (WK 1)</b>	<b>Key Vocabulary</b>	<b>Homework</b>
<p>how to construct the balanced formula of an ionic compound using ion charges; how to write balanced symbol equations, with state symbols, for the metals listed in the reactivity series reacting with oxygen, water, and acid.</p>		Ion	PPQ based on Ionic Compounds
<b>LP3 RLW, I will:</b>	<b>24/02/25 - (WK 2)</b>		
<p>review my learning, recalling and applying key knowledge, and focus on closing any gaps in my knowledge.</p>			
<b>In LP3.7, I will know:</b>	<b>03/03/25 - (WK 1)</b>	<b>Key Vocabulary</b>	<b>Homework</b>
<p>how to describe displacement reactions using ionic equations. Extended Task.</p>		Displacement	Retrieval Questions based key misconceptions from LP3
<b>Resources to support learning:</b>			
Resource booklet, Knowledge organiser, BBC GCSE Bitesize, Free GCSE Science videos on YOUTUBE. COGNITO Science			
<b>FFET Award Challenge for this Learning Programme:</b>			
LP3 Year 10 Science: Take part in an extra curricular STEM challenge			

PRT Task 1

PRT Task 2