

## Year 13 Chemistry EDN Learning Programme 2

<p>The LORIC skill focus for his LP is: ORGANISATION The Moral Virtues focus for this LP are: COMPASSION and HONESTY Compassion - the quality of feeling pity and concern for the sufferings or misfortunes of others. Honesty - the quality of being truthful.</p>		<p><b>Literacy:</b></p> <ul style="list-style-type: none"> <li>• Capital letters must be used at the start of sentences and for the first letter of proper nouns</li> <li>• Full stops must be used at the end of a sentence</li> <li>• Question marks must be used at the end of a question</li> <li>• Apostrophes should only be used for possession or omission</li> <li>• Days of the week and months must be spelled correctly</li> <li>• Key words must be spelled correctly</li> </ul>	
<p><b>What will I be learning about in this Learning Programme?</b> Equilibrium constants <math>K_c</math> and <math>K_p</math> and how to control the position of equilibrium.</p>			
<p><b>Where have I seen this learning before?</b> GCSE chemistry rates of reaction, A level chemistry module 3, including Le Chatelier's Principle.</p>			
<p><b>What could I use it for?</b> Degree level chemistry, industrial processes and developing medications.</p>			
<b>In LP2.1, I will know:</b>	<b>21/10/24 - (WK 2)</b>	<b>Key Vocabulary</b>	<b>Homework</b>
<p>how to determine the quantities of reactant and product at equilibrium; how to calculate <math>K_c</math> and determine units.</p>		Equilibrium	PPQ on $K_c$
<b>In LP2.2, I will know:</b>	<b>04/11/24 - (WK 1)</b>	<b>Key Vocabulary</b>	<b>Homework</b>
<p>how to calculate <math>K_p</math> and determine units; how to apply <math>K_c</math> and/or <math>K_p</math> to other equilibrium constants.</p>		Mole fraction	PPQ of $K_p$
<b>LP2 RLW, I will:</b>	<b>11/11/24 - (WK 2)</b>		
<p>review my learning, recalling and applying key knowledge, and focus on closing any gaps in my knowledge.</p>			
<b>In LP2.3, I will know:</b>	<b>18/11/24 - (WK 1)</b>	<b>Key Vocabulary</b>	<b>Homework</b>
<p>how to predict how changes within a system affect the position of equilibrium; how to determine the lattice enthalpy of formation.</p>		Enthalpy	PPQ on enthalpy changes
<p>Extended Task.</p>			
<b>In LP2.4, I will know:</b>	<b>25/11/24 - (WK 2)</b>	<b>Key Vocabulary</b>	<b>Homework</b>
<p>how to construct a Born-Haber cycle; how to complete a summative assessment.</p>		Born Haber Cycle	PPQ on Born Haber cycles
<b>In LP2.5, I will know:</b>	<b>02/12/24 - (WK 1)</b>	<b>Key Vocabulary</b>	<b>Homework</b>
<p>how to determine enthalpy changes in solutions; how to determine enthalpy change of hydration.</p>		Electron affinity	PPQ on enthalpy changes
<b>In LP2.6, I will know:</b>	<b>09/12/24 - (WK 2)</b>	<b>Key Vocabulary</b>	<b>Homework</b>
<p>how to describe and explain the factors affecting lattice enthalpy; how to calculate entropy changes.</p>		Entropy	PPQ on entropy
<p>Extended Task.</p>			
<b>In LP2.7, I will know:</b>	<b>16/12/24 - (WK 1)</b>	<b>Key Vocabulary</b>	<b>Homework</b>
<p>how to use the Gibb's free energy equation; how to use free energy to predict feasibility.</p>		Gibb's energy	PPQ on Gibb's energy
<p><b>Resources to support learning:</b> OCR A level chemistry text book, MaChem Guy.</p>			
<p><b>FFET Award Challenge for this Learning Programme:</b> Complete a full paper 1 past paper.</p>			

